

Types of Reactions

Predicting Product

Practice

Please note that the following reactions are NOT necessarily balanced! We are just focusing on types of reactions and predicting products. You can balance them later!

Things to Remember – *crossing over to make neutral compounds, diatomics, roman numerals, rewriting water as $H(OH)$, write good formulas and then balance at the end to fix your numbers!*

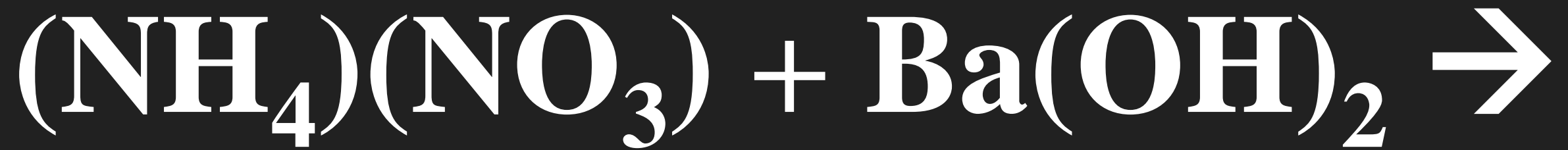
Page Set Up

Q#	Type	Reaction
1		
2		
3		
Etc...		



NaBr →

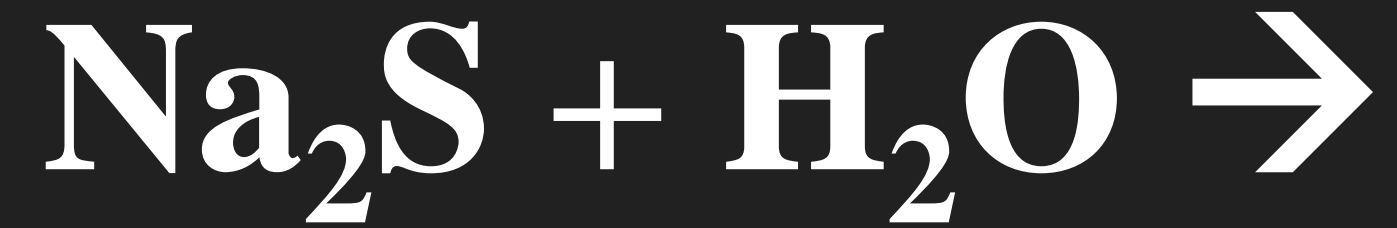


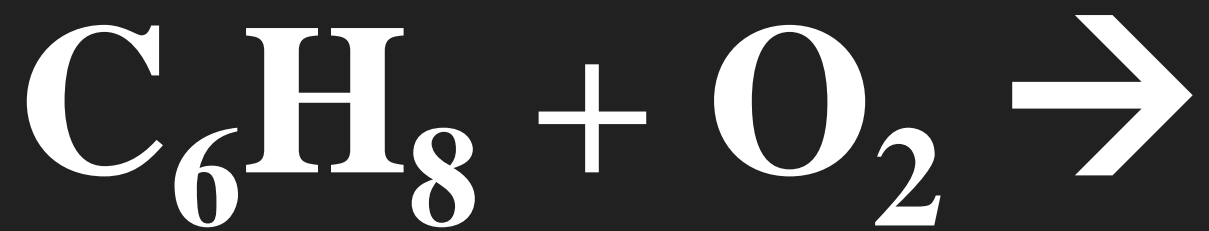










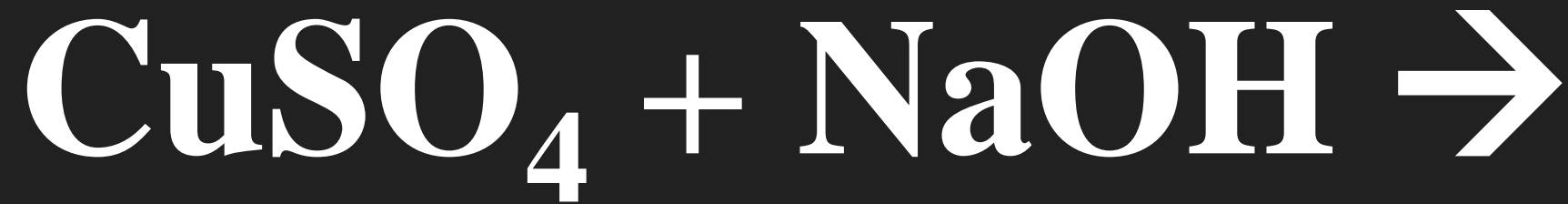








assume copper (II) in your product



assume copper (I) is in your product







assume iron (IV) is in your product































assume nickle(I) is in your product

$\text{CuO} \rightarrow$



